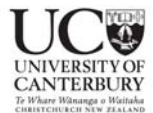


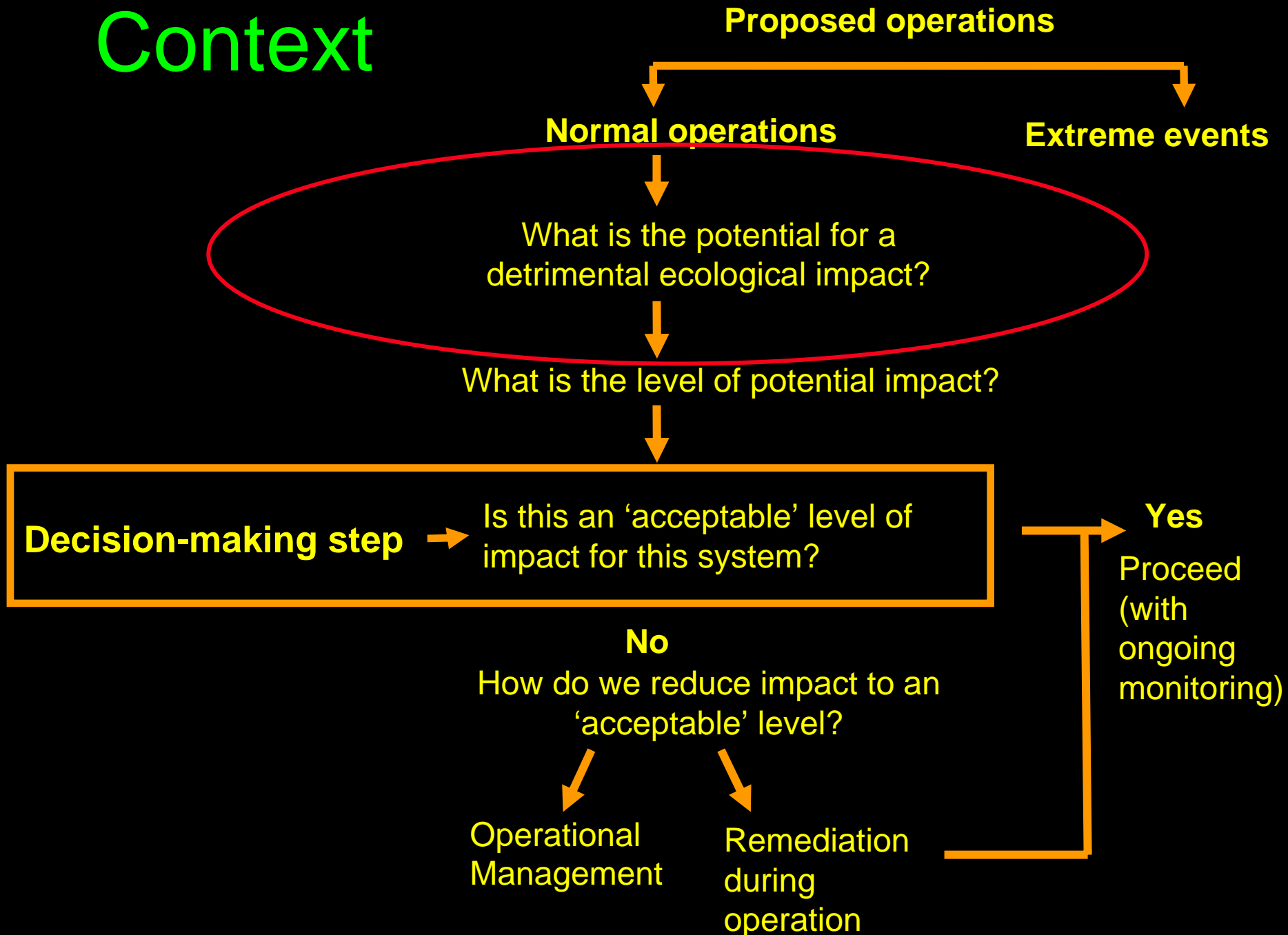
Arsenic discharge from an historic gold mine site, Waiuta, Westland

Laura Haffert, Dave Craw and James Pope

Mine drainage framework



Context



Introduction

- Part of a PhD project
 - Focus: Arsenic mobility at historic mine sites
- Close association of the arsenic and gold
concentrating gold = concentrating arsenic
- Arsenic-rich processing residues
unrestricted disposal at historic sites

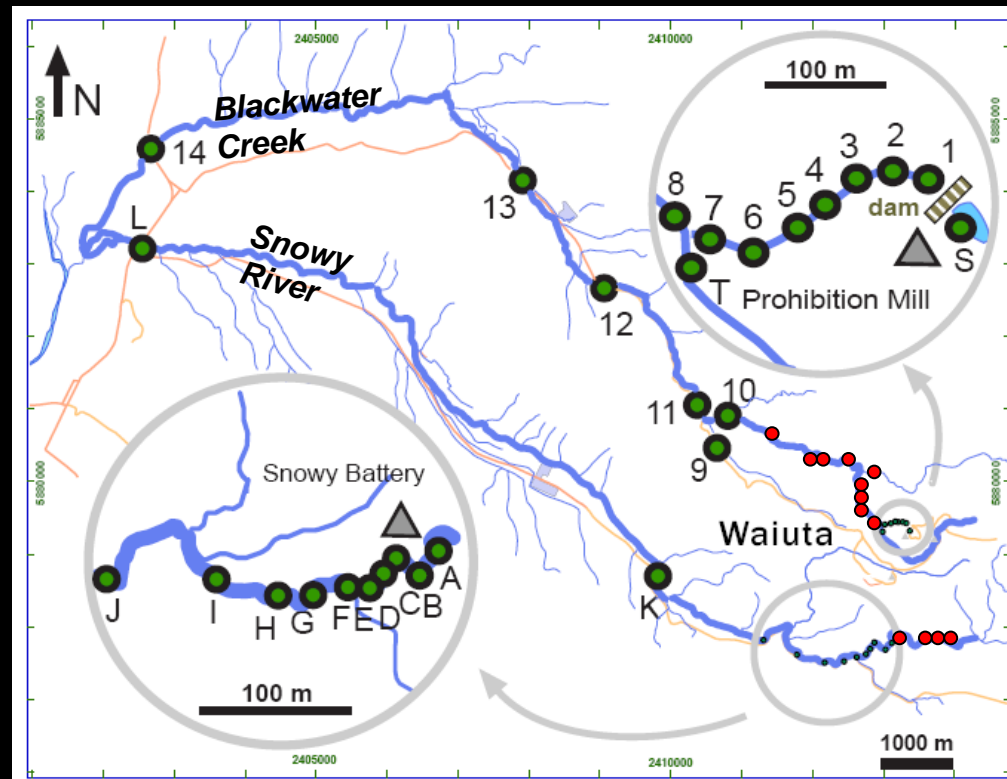
Introduction



- Hard rock gold deposits
- Host rock: Greenland Group metasediments (contain carbonates)
- Prohibition Mill site: 1938 – 1951
- Arsenopyrite-rich ore

Methodology - sampling

- A range of solid samples from processing plant, substrate and wetland
- Water samples:



Methodology - analysis

Method

Purpose

Solid samples

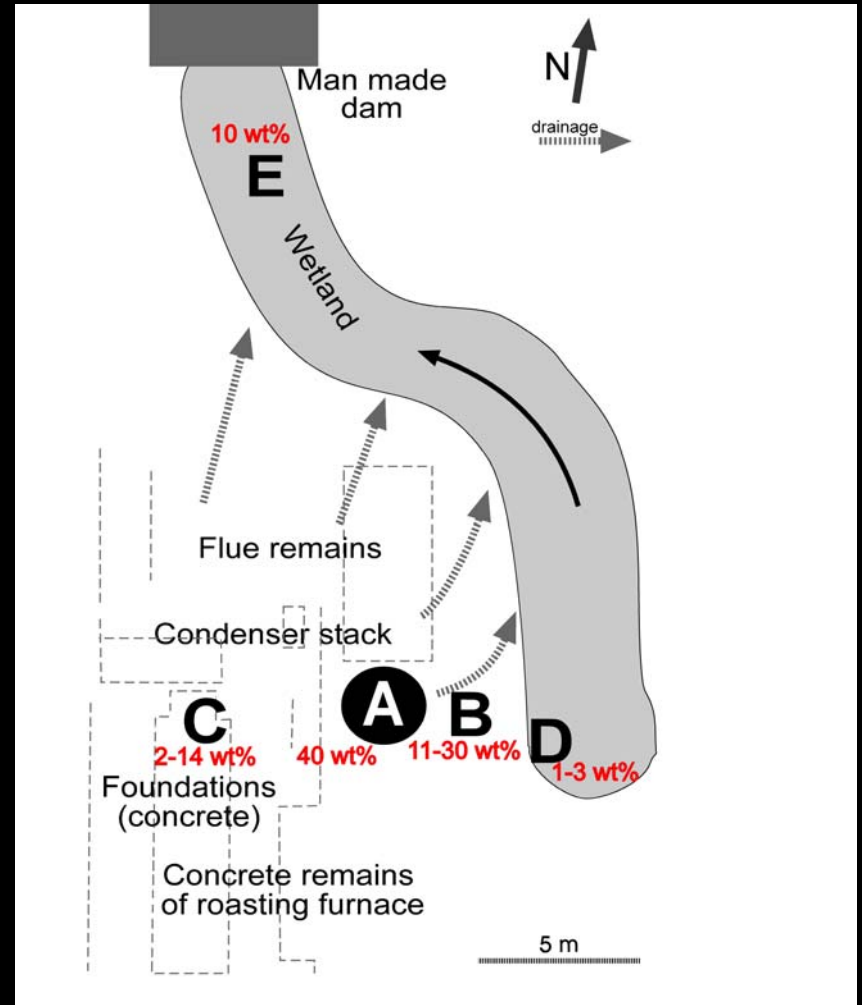
- ICP-MS Total arsenic concentration
- XRD Mineral identification
- Microprobe Micro-scale imaging for arsenic phase characterisation
- Handheld XRF Arsenic distribution and extent in substrate

Water samples

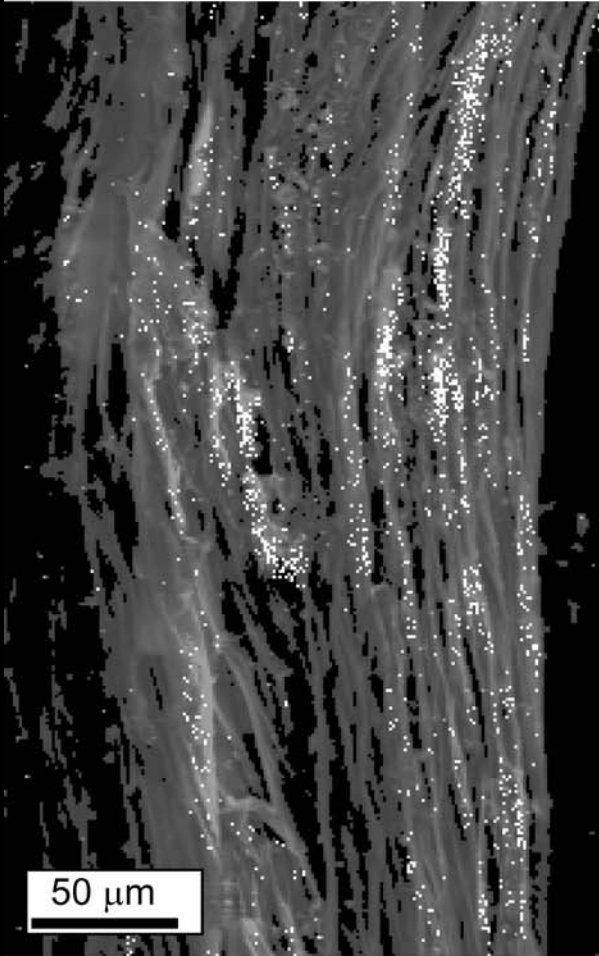
- Filtering (< 0.45 μm) Separation of solid from dissolved arsenic
- ICP-MS analysis Dissolved arsenic concentration
- Major ion profile General water quality
- pH (in situ)

Selected results

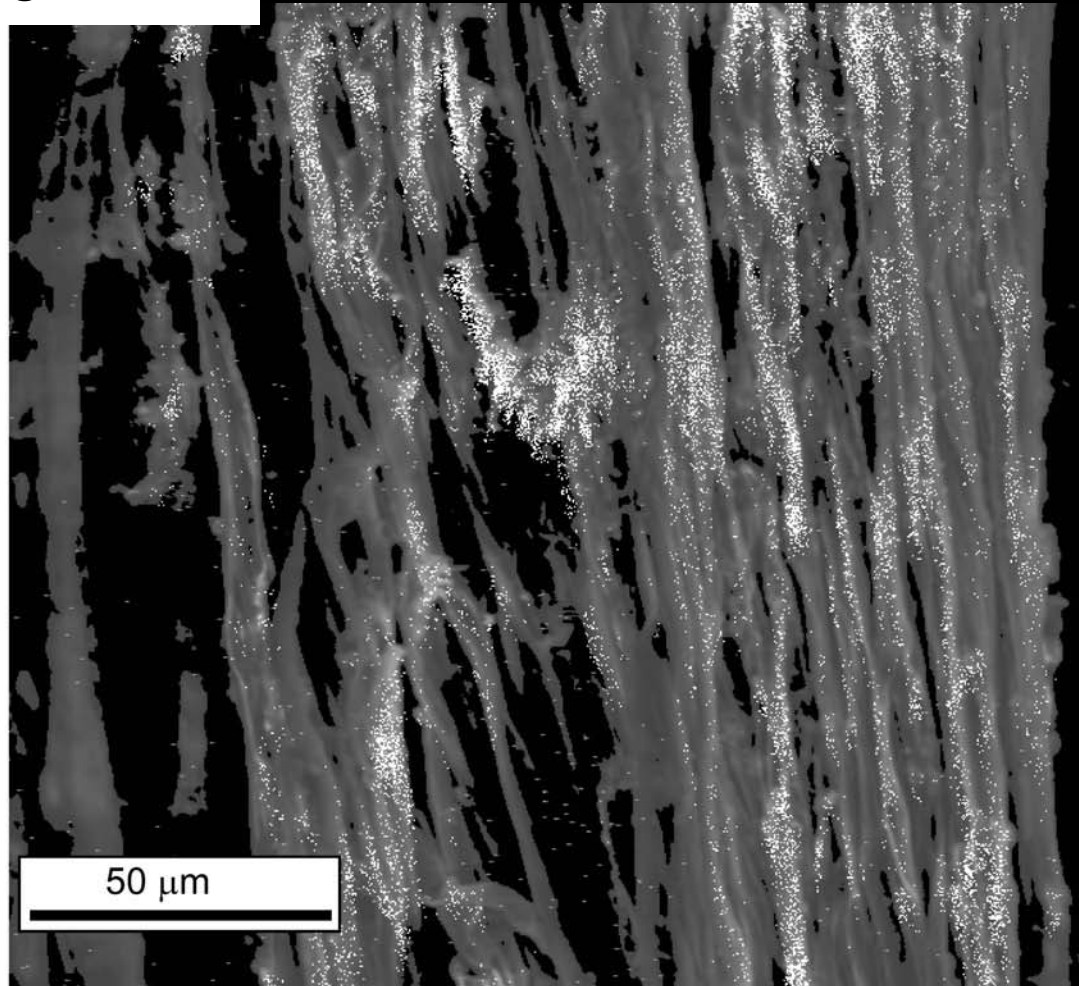
Arsenic in solids: concentrations



Moss, *Pohlia wahlenbergii*
Microprobe images



Moss As:
0.8-3.1 wt%



Water As:
50-77 mg/L

Substrate As:
3-16 wt%

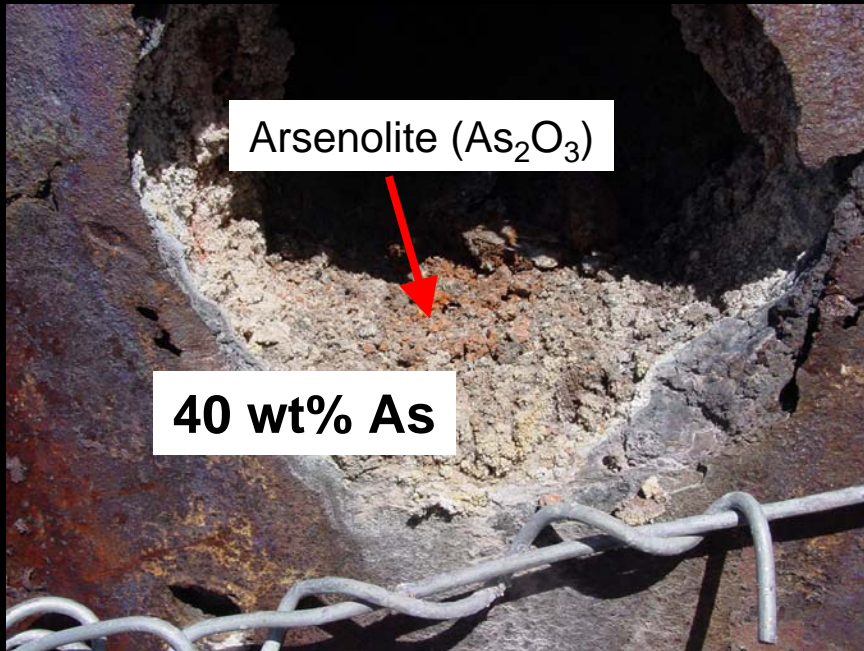
Intake of ca. 0.2 g of this
material could be fatal
→ Human access has to be
prevented



Selected results

Arsenic in solids: *mineralogy*

Roaster



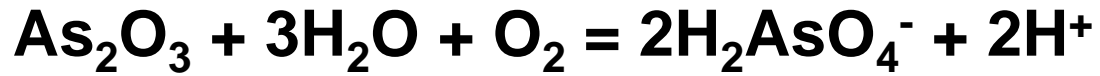
Arsenolite - roasting by-product
Occurrence: roaster and its vicinity, wetland
(below surface layer)

Substrate



Scorodite - common secondary arsenic mineral
Occurrence: mainly as substrate cement

Oxidation causes acidification (not from pyrite):



Arsenolite
As₂O₃

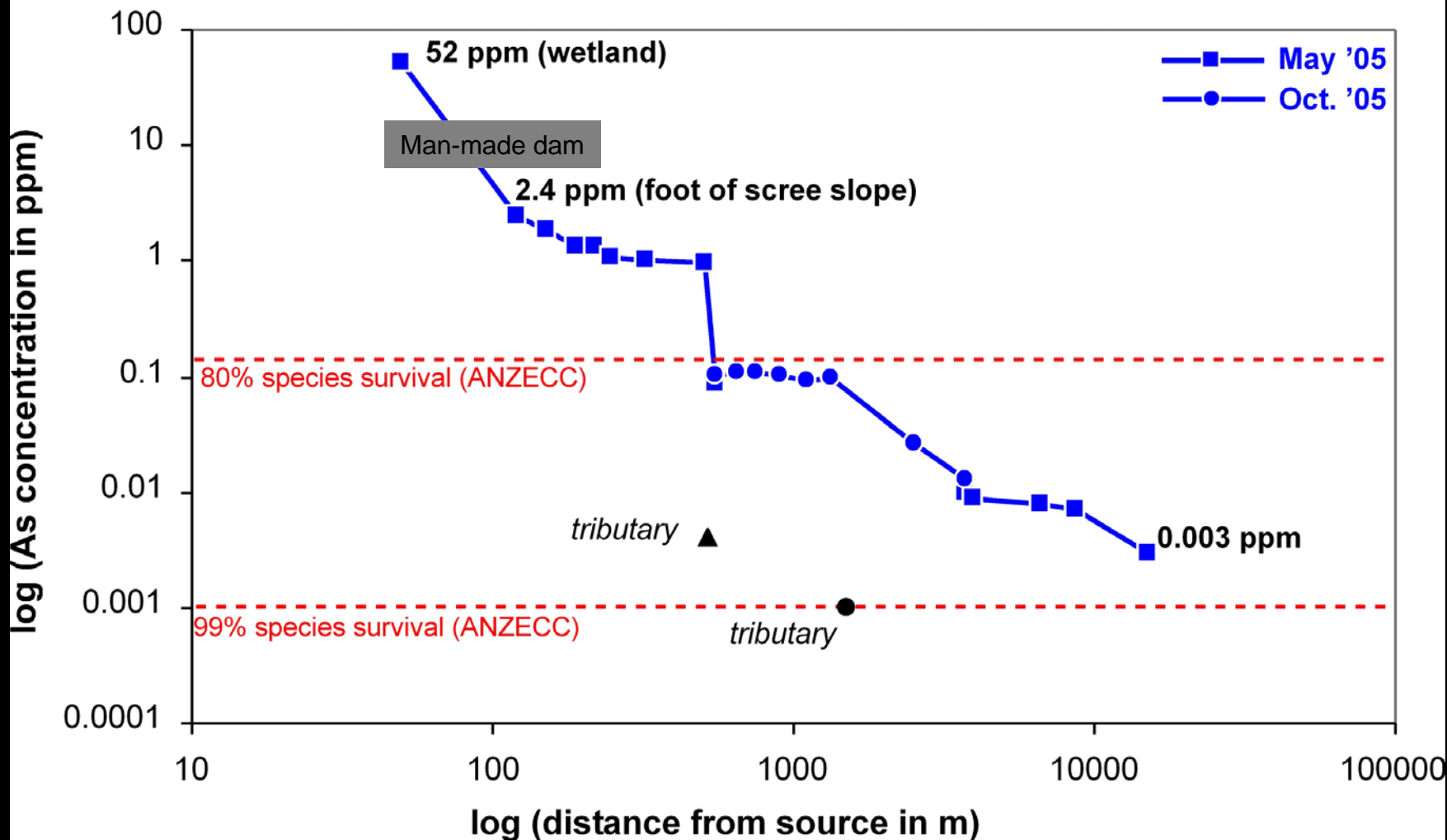
pH down to 3

**Dissolved arsenic runoff
forms scorodite cement**

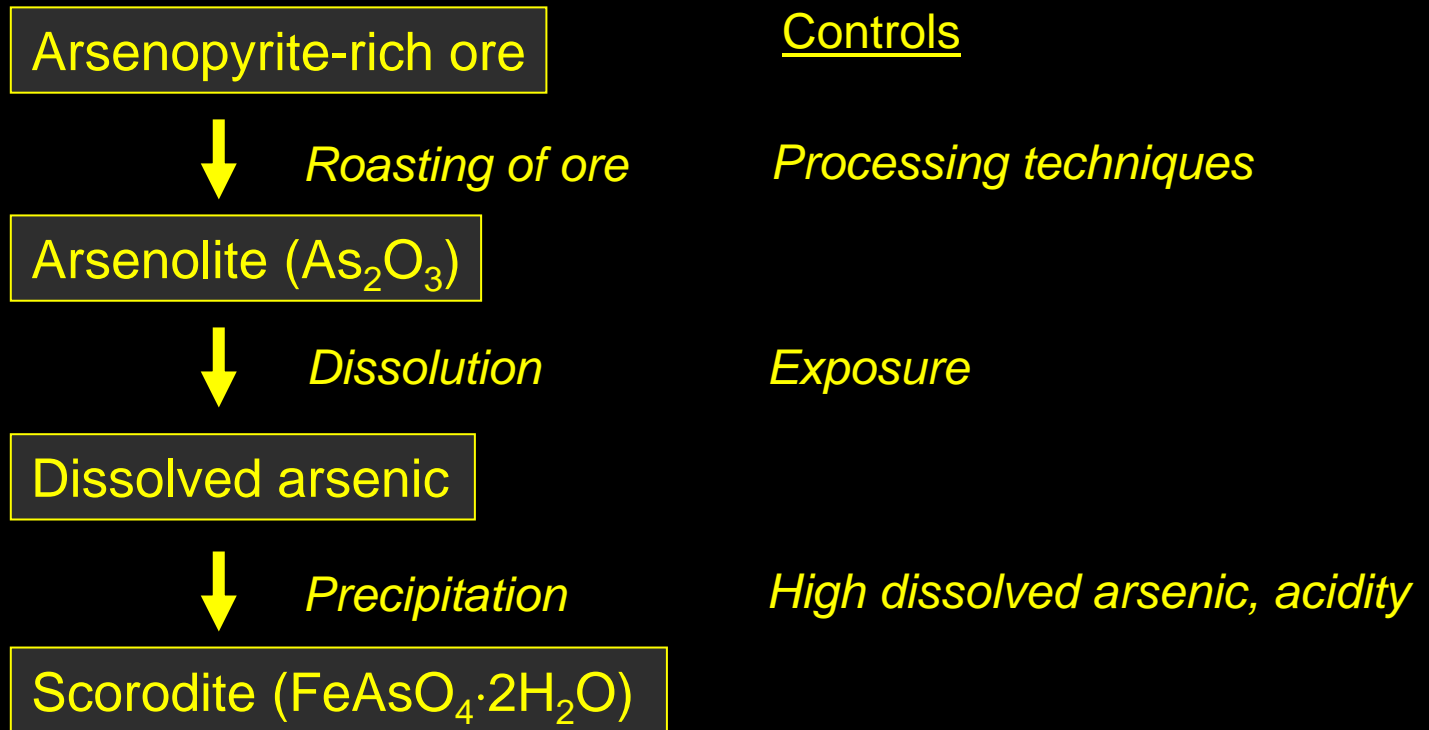
Selected results

Arsenic in water:

Prohibition Mill site drainage *dissolved arsenic*



Local controls on site impact



Large volume of arsenic temporarily immobilised as scorodite

Management perspective: Removal of arsenolite lowers dissolved arsenic concentrations and increases pH
→ Scorodite becomes unstable → remobilisation of arsenic

Local controls on site impact

Processes in the man-made dam

Wetland

- High dissolved arsenic (52 mg/L)
- Acidic (pH 3-4, from arsenolite oxidation)
- Carbonate deficient

Man-made dam:
Greenland Group boulders

- Dissolution of Fe-bearing carbonates:
 - Dissolved Fe
 - Neutralisation
- Precipitation of Fe (insoluble at circum-neutral pH) as iron oxyhydroxide (HFO)
 - Adsorption of arsenic onto HFO (passive treatment)

Creek

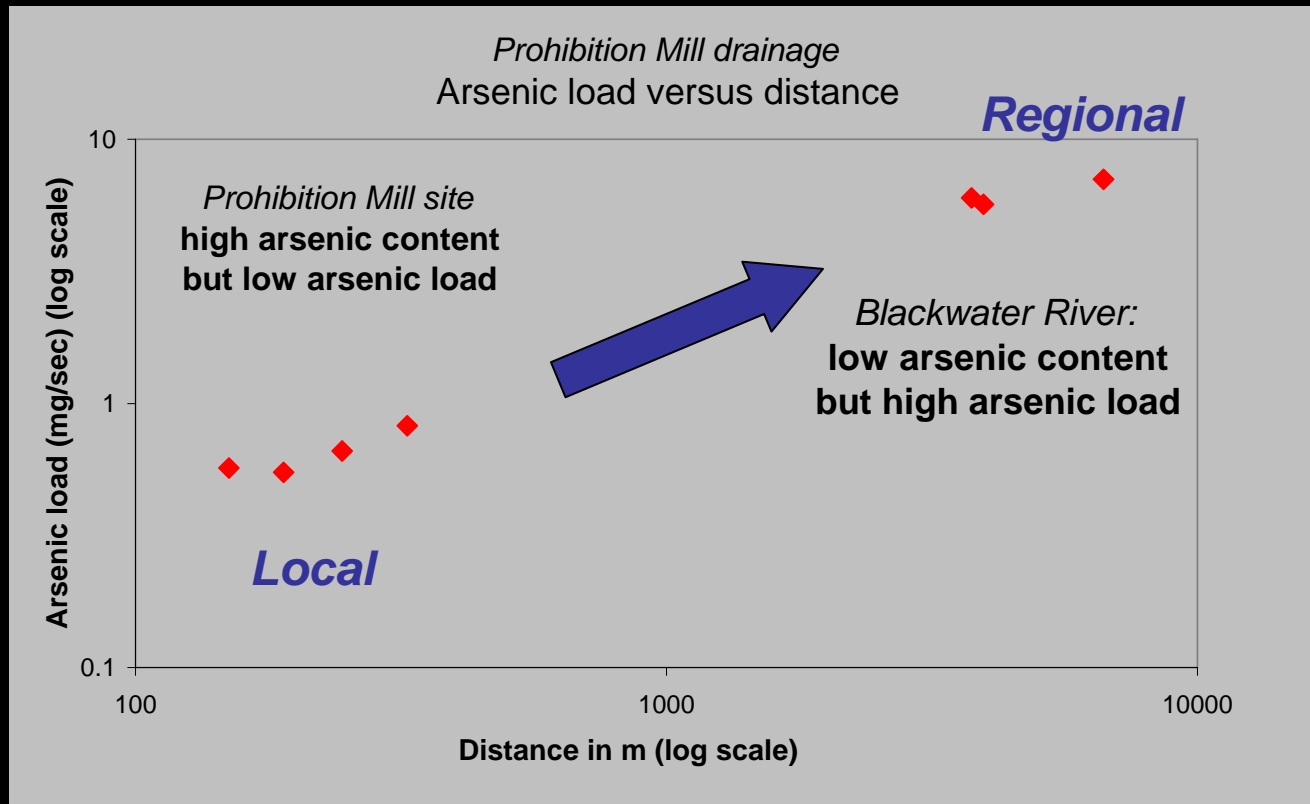
- An order of magnitude less arsenic than in wetland (2 mg/L)

Management perspective: No addition of lime or local rocks!

- Increased pH increases solubility of scorodite
- Increased dissolved carbonate

Regional controls on site impact

- Regional attenuation is via dilution (less effective)
- Prohibition mill site contributions - regional background contributions (natural):
Arsenic load (mg/sec) = flow rate (L/sec) x arsenic concentration (mg/L)



Management perspective: site clean up will not have a significant influence on downstream arsenic concentrations

Conclusions

Site impact

Local:

- Prohibition mill site one of the most toxic sites in South Island:

very arsenic-rich residues (up to 40 wt%), some in the form of very soluble arsenolite

→ site unsuitable for human access

→ very disrupted site ecosystem

High dissolved arsenic in wetland (50 mg/L) and creek (2 mg/l)

→ Strongly disrupted stream biota several hundred metres downstream

Regional:

- Site impact is negligible on a regional scale

Conclusions

Controls on site impact

Local:

- Arsenic mineralogy and their stabilities (dynamic system)
 - Effective attenuation in man-made dam
(sensitive to water quality and pH)
-

Regional:

- Dilution to elevated natural background

Conclusion

Management perspective:

- presently the site is acidic, carbonate deficient with very high dissolved arsenic concentrations from arsenolite dissolution.

Changes of any of these parameters can result in :

- Remobilisation of arsenic through scorodite dissolution
- Reduction or prevention of efficient attenuation in dam



Site remediation options should be based on a geochemical understanding of the site!

-
- Site remediation will not change water quality of the downstream environment on a regional scale

Acknowledgements

Thanks to

- DoC for access and logistical support
- FRST via CRL for funding
- And to all the adventurous field assistants for helping out on such a toxic site.