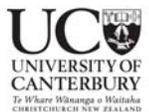


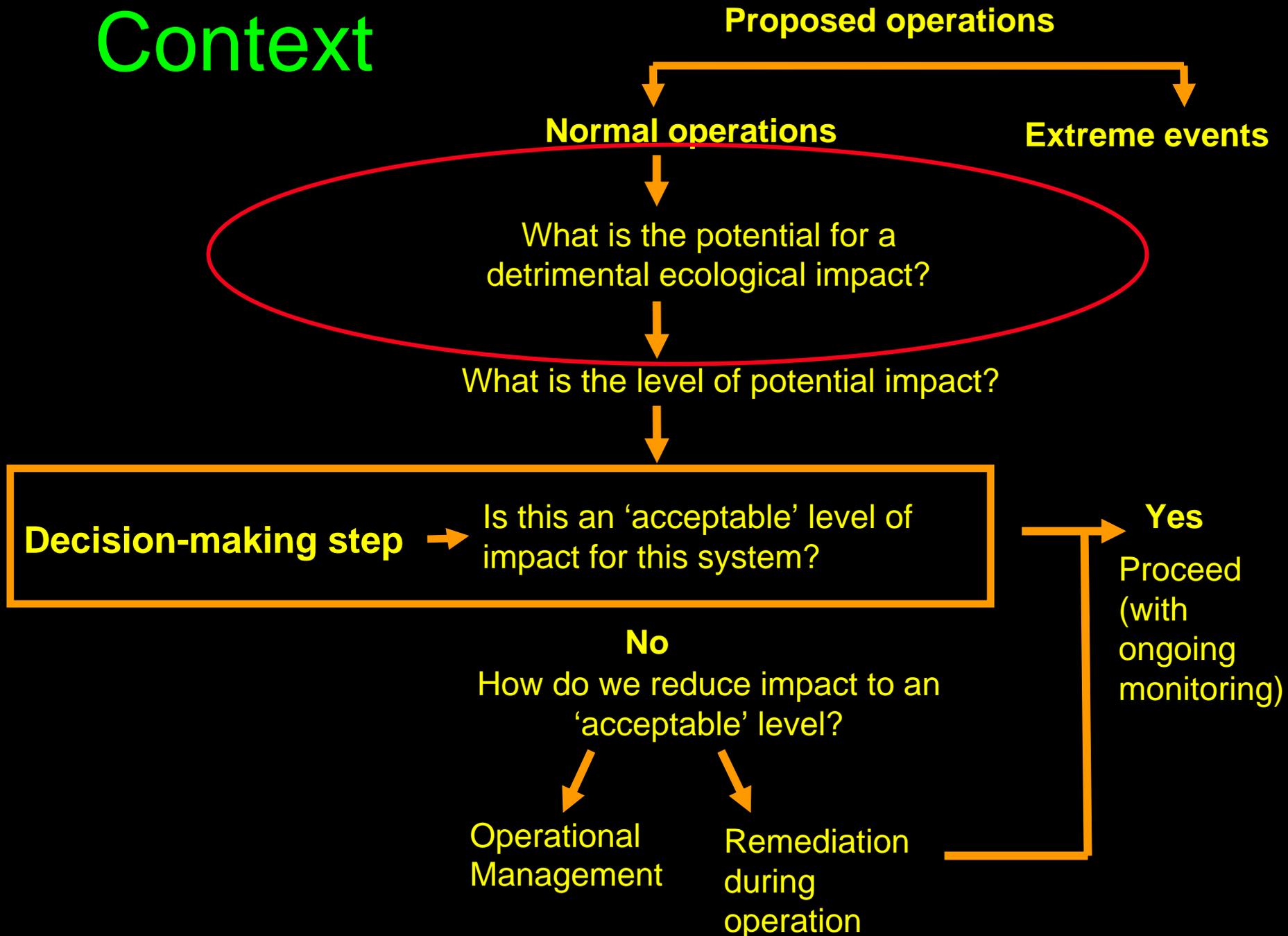
# Arsenic discharge from an historic gold mine site, Waiuta, Westland

Laura Haffert, Dave Craw and James Pope

Mine drainage framework



# Context



# Introduction

- Part of a PhD project
  - Focus: Arsenic mobility at historic mine sites
- Close association of the arsenic and gold  
concentrating gold = concentrating arsenic
- Arsenic-rich processing residues  
unrestricted disposal at historic sites

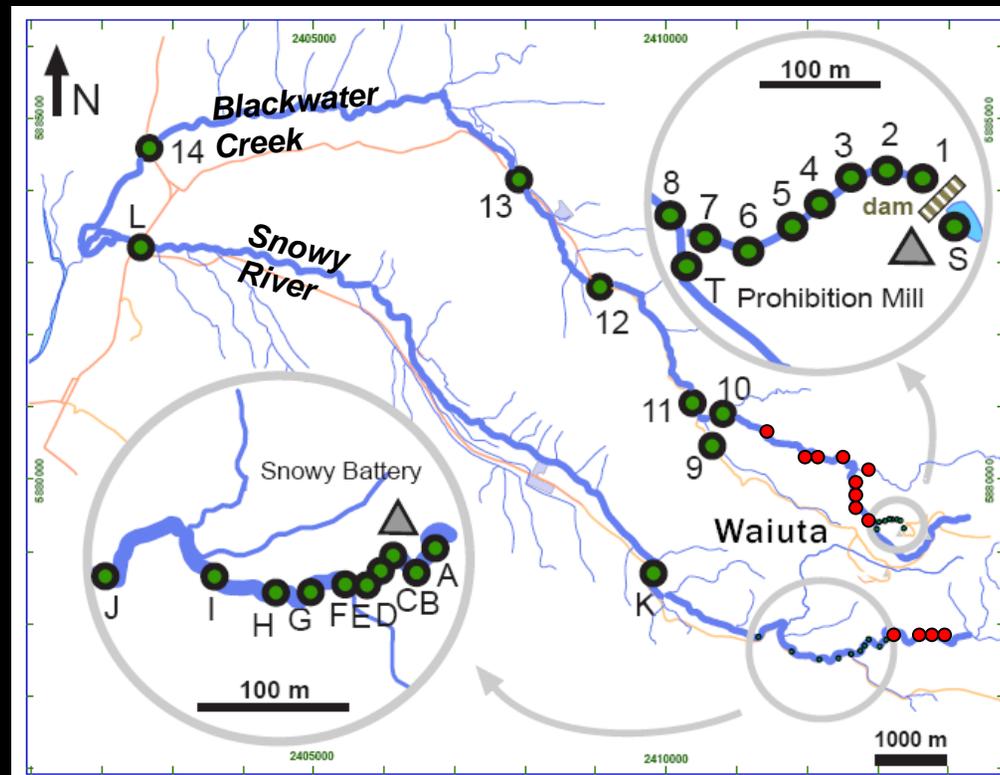
# Introduction



- Hard rock gold deposits
- Host rock: Greenland Group metasediments (contain carbonates)
- Prohibition Mill site: 1938 – 1951
- Arsenopyrite-rich ore

# Methodology - sampling

- A range of solid samples from processing plant, substrate and wetland
- Water samples:



# Methodology - analysis

## Method

## Purpose

### Solid samples

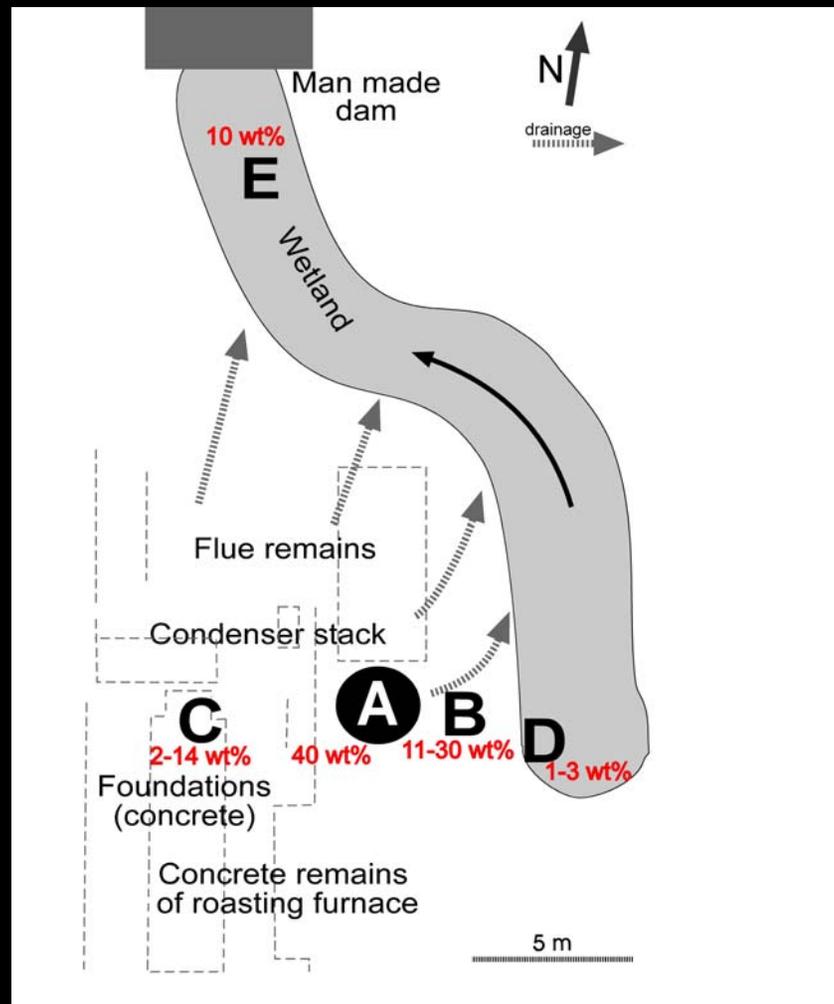
- ICP-MS Total arsenic concentration
- XRD Mineral identification
- Microprobe Micro-scale imaging for arsenic phase characterisation
- Handheld XRF Arsenic distribution and extent in substrate

### Water samples

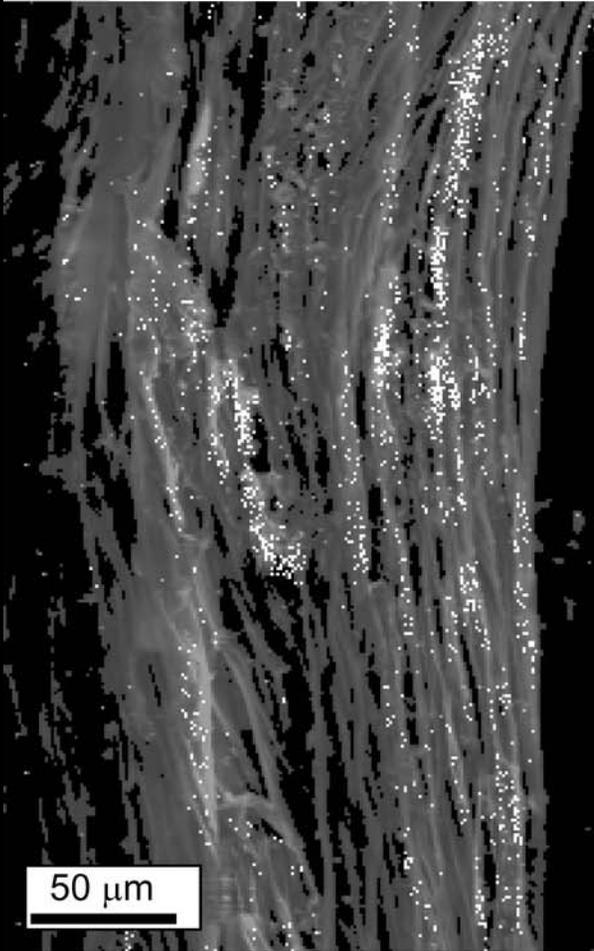
- Filtering (< 0.45  $\mu\text{m}$ ) Separation of solid from dissolved arsenic
- ICP-MS analysis Dissolved arsenic concentration
- Major ion profile General water quality
- pH (in situ)

# Selected results

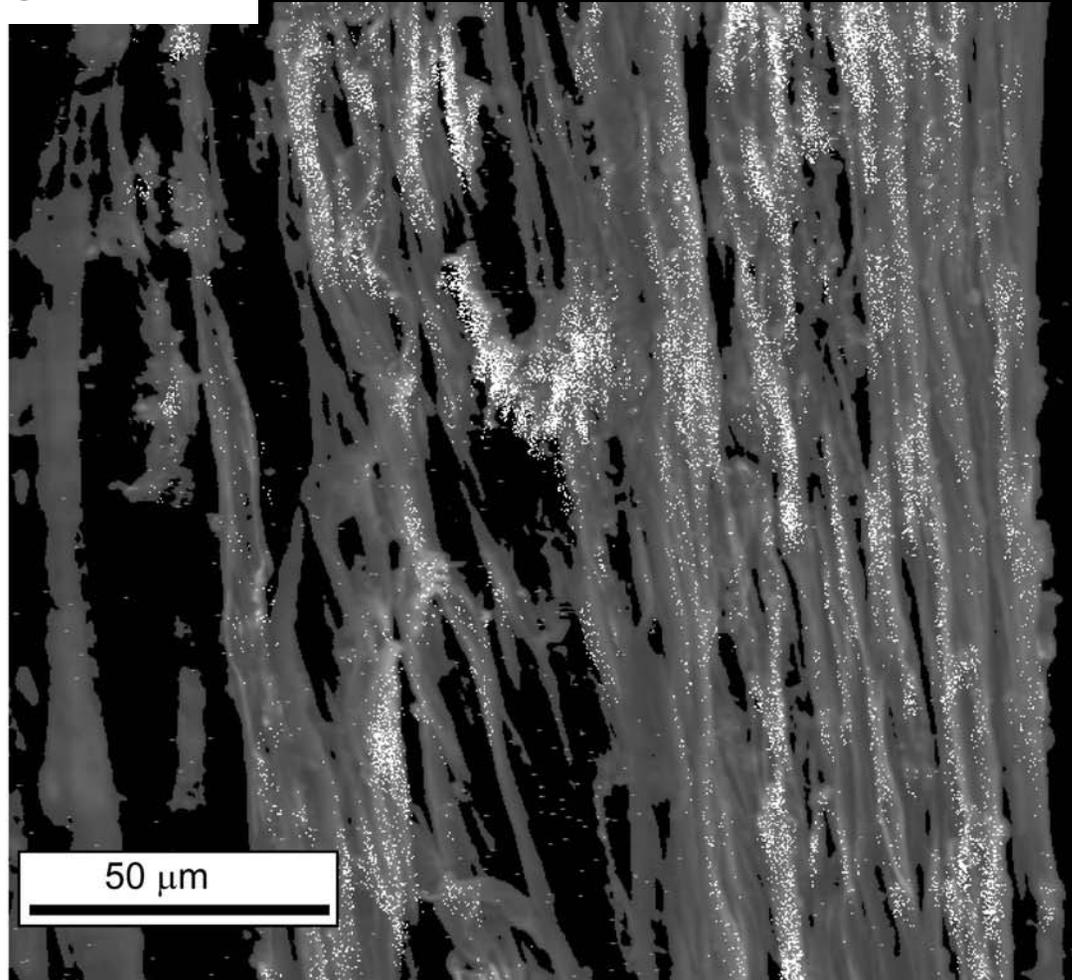
## Arsenic in solids: concentrations



Moss, *Pohlia wahlenbergii*  
Microprobe images



**Moss As:**  
**0.8-3.1 wt%**



**Water As:**  
**50-77 mg/L**

**Substrate As:**  
**3-16 wt%**

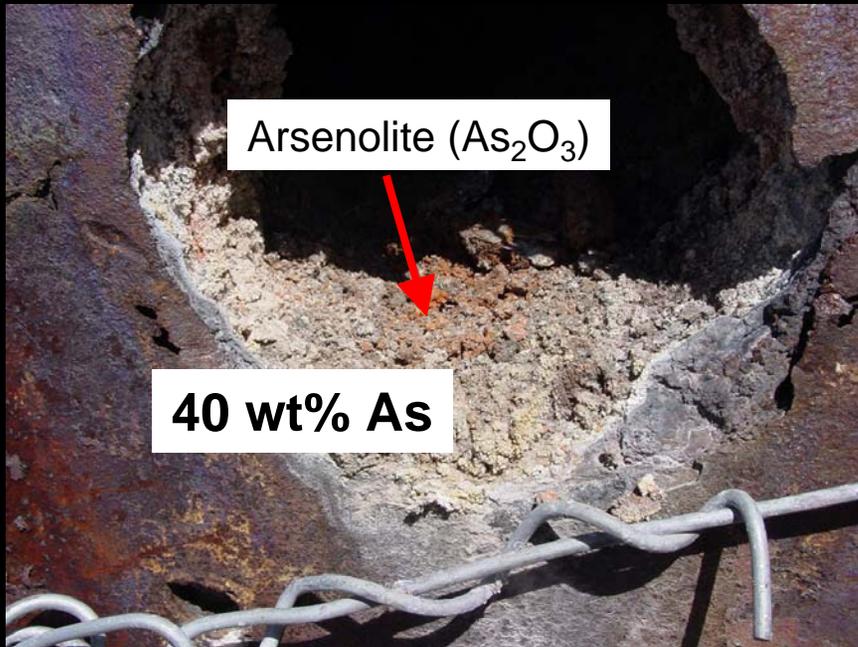
Intake of ca. 0.2 g of this  
material could be fatal  
→ Human access has to be  
prevented



# Selected results

## Arsenic in solids: *mineralogy*

### Roaster



**Arsenolite** - roasting by-product  
Occurrence: roaster and its vicinity, wetland  
(below surface layer)

### Substrate



**Scorodite** - common secondary arsenic mineral  
Occurrence: mainly as substrate cement

**Oxidation causes acidification (not from pyrite):**



**Arsenolite**  
**As<sub>2</sub>O<sub>3</sub>**

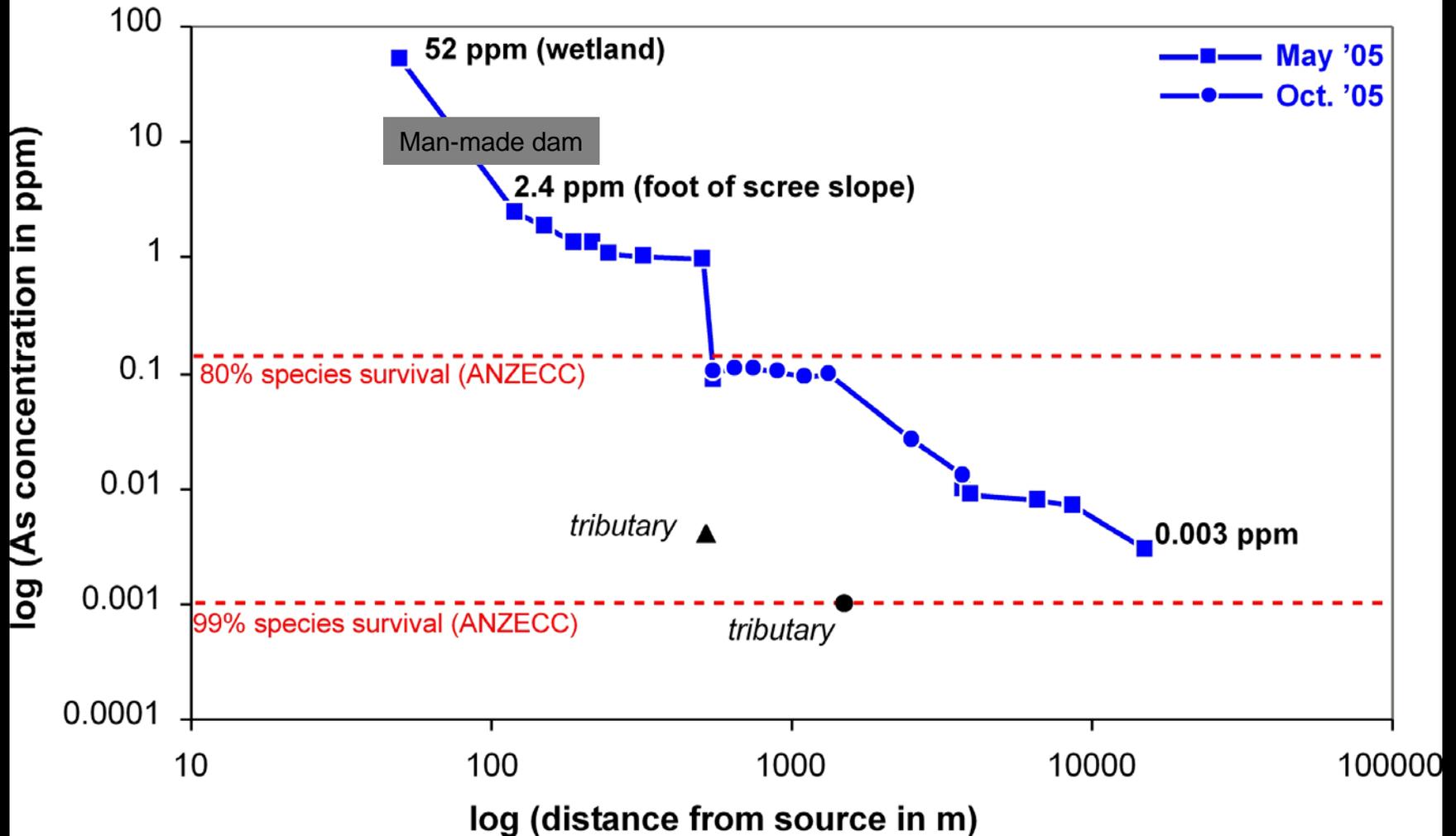
**pH down to 3**

**Dissolved arsenic runoff  
forms scorodite cement**

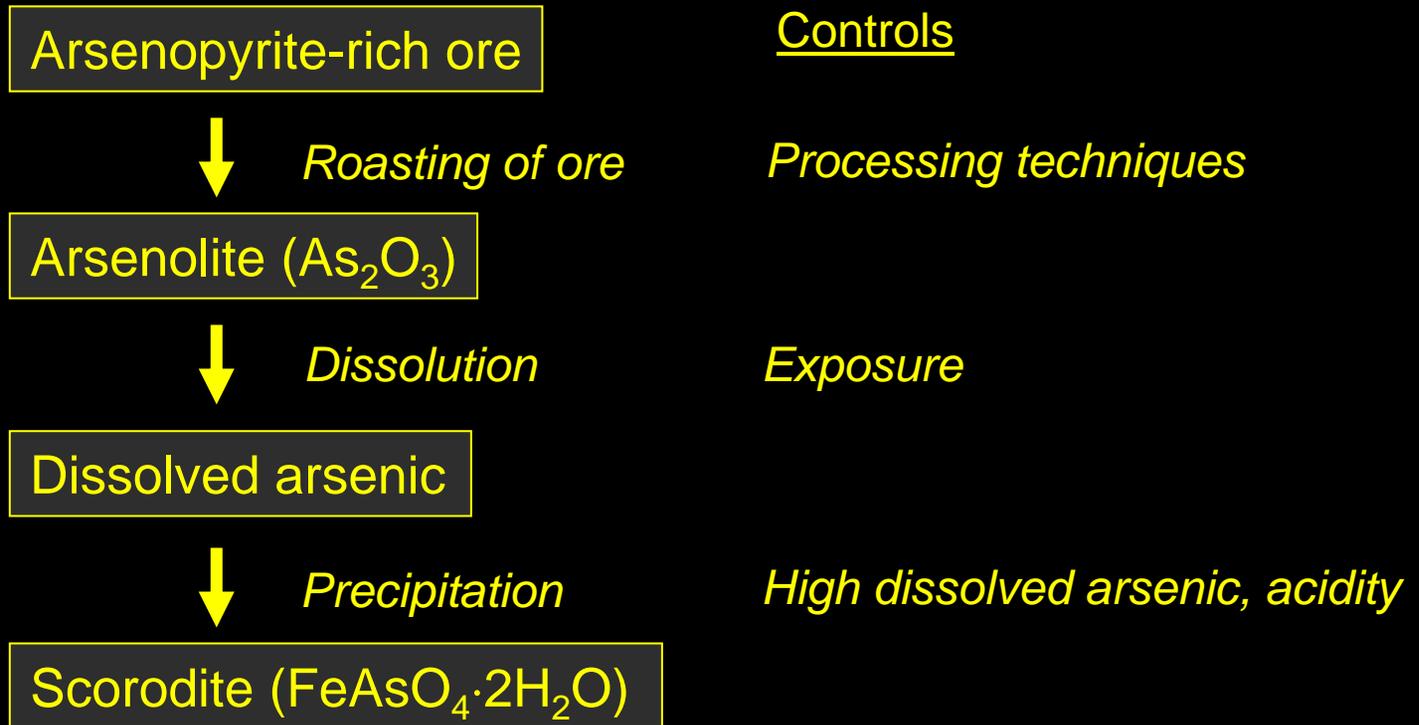
# Selected results

## Arsenic in water:

### Prohibition Mill site drainage *dissolved arsenic*



# Local controls on site impact



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Large volume of arsenic temporarily immobilised as scorodite

**Management perspective:** Removal of arsenolite lowers dissolved arsenic concentrations and increases pH  
→ Scorodite becomes unstable → remobilisation of arsenic

# Local controls on site impact

## *Processes in the man-made dam*

Wetland

- High dissolved arsenic (52 mg/L)
- Acidic (pH 3-4, from arsenolite oxidation)
- Carbonate deficient

Man-made dam:  
Greenland Group boulders

- Dissolution of Fe-bearing carbonates:
  - Dissolved Fe
  - Neutralisation
- Precipitation of Fe (insoluble at circum-neutral pH) as iron oxyhydroxide (HFO)
  - Adsorption of arsenic onto HFO (passive treatment)

Creek

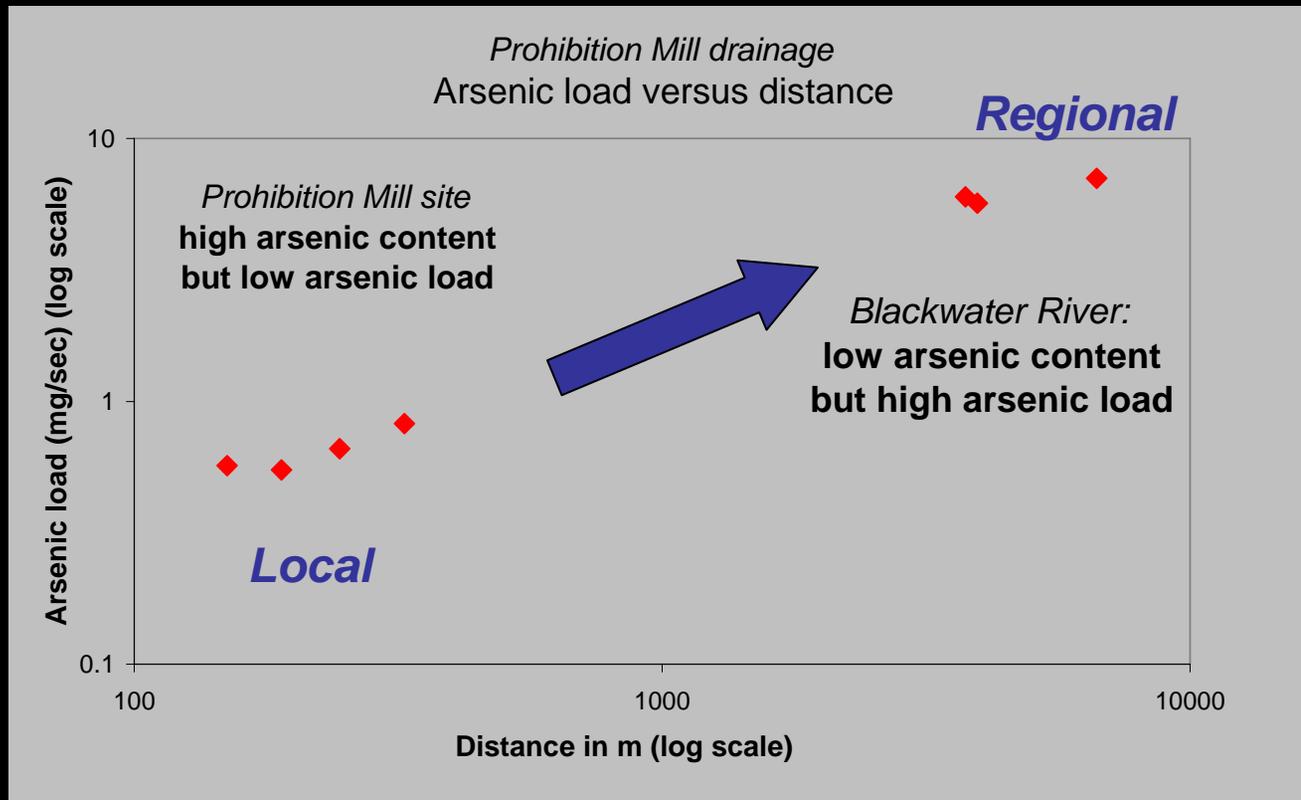
- An order of magnitude less arsenic than in wetland (2 mg/L)

**Management perspective:** No addition of lime or local rocks!

- Increased pH increases solubility of scorodite
- Increased dissolved carbonate

# Regional controls on site impact

- Regional attenuation is via dilution (less effective)
- Prohibition mill site contributions - regional background contributions (natural):  
**Arsenic load (mg/sec) = flow rate (L/sec) x arsenic concentration (mg/L)**



**Management perspective: site clean up will not have a significant influence on downstream arsenic concentrations**

# Conclusions

## Site impact

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### *Local:*

- Prohibition mill site one of the most toxic sites in South Island:

very arsenic-rich residues (up to 40 wt%), some in the form of very soluble arsenolite

→ site unsuitable for human access

→ very disrupted site ecosystem

High dissolved arsenic in wetland (50 mg/L) and creek (2 mg/l)

→ Strongly disrupted stream biota several hundred metres downstream

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### *Regional:*

- Site impact is negligible on a regional scale

# Conclusions

## Controls on site impact

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### *Local:*

- Arsenic mineralogy and their stabilities (dynamic system)
  - Effective attenuation in man-made dam  
(sensitive to water quality and pH)
- 

### *Regional:*

- Dilution to elevated natural background

# Conclusion

## *Management perspective:*

- presently the site is acidic, carbonate deficient with very high dissolved arsenic concentrations from arsenolite dissolution.

Changes of any of these parameters can result in :

- Remobilisation of arsenic through scorodite dissolution
- Reduction or prevention of efficient attenuation in dam



Site remediation options should be based on a geochemical understanding of the site!

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- Site remediation will not change water quality of the downstream environment on a regional scale

# Acknowledgements

Thanks to

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- And to all the adventurous field assistants for helping out on such a toxic site.